612 Final review

Jinchen He

Heat capacity

$$n = g \int \frac{dk}{2\pi^2} k^2 \cdot n_f \left(\frac{\epsilon - \mu}{T} \right)$$

With steep descent,

$$\frac{u}{n} = \frac{u_0 + \Delta u}{n_0 + \Delta n}$$

$$\Delta n = g \sum_{n=0}^{\infty} \frac{f^{(n)}(\epsilon)}{n!} \Big|_{\epsilon = \mu} \cdot T^{n+1} C_n$$

$$\Delta u = g \sum_{n=0}^{\infty} \frac{(\epsilon \cdot f)^{(n)}(\epsilon)}{n!} \Big|_{\epsilon = \mu} \cdot T^{n+1} C_n$$

$$C_n = \int dy \ y^n \cdot \Delta(y)$$

$$f(\epsilon) = \frac{(2m)^{3/2}}{4\pi^2} \epsilon^{1/2}$$

we got

$$\frac{u}{n} = \frac{U}{N} = \frac{3}{5}\epsilon_f \left(1 + \frac{5}{12}\pi^2 \left(\frac{T}{\epsilon_f} \right)^2 + \dots \right)$$

$$C_V = \frac{dU}{dT} \Big|_{N,V} = \frac{N\pi^2}{2\epsilon_f} T + O(T^2)$$

Lattice

$$\frac{1}{2}\log\left(\frac{1+x}{1-x}\right) = \tanh^{-1}(x)$$

Critial exponent

Specific heat C, order parameter Ψ , susceptibility χ and correlation length ξ .

With
$$t = \frac{T - T_c}{T}$$
, for $t > t_0$

$$C \sim t^{-\alpha}$$
, $\Psi \sim 0$, $\chi \sim t^{-\gamma}$, $\xi \sim t^{-\nu}$

for $t < t_0$

$$C \sim (-t)^{-\alpha'}, \ \Psi \sim (-t)^{-\beta}, \ \chi \sim (-t)^{-\gamma'}, \ \xi \sim (-t)^{-\nu'}$$

where α , α' , β , γ , $\gamma' > 0$.

In ising model we have $\Psi=\langle\sigma\rangle$ and $\chi=\frac{\partial\sigma}{\partial h}\Big|_{h=0}$.

Two laws,

$$\alpha + 2\beta + \gamma = 2$$

$$v = \frac{2 - \alpha}{d}, \text{ where d is dimension of space}$$

Bubble

If we have a bubble of phase II with free energy density f_2 in the sea of phase I with free energy density f_1 , and $f_2 < f_1$.

Then the free energy difference is

$$\Delta F = V(f_2 - f_1) + \sigma A = \frac{4\pi}{3}R^3(f_2 - f_1) + 4\pi R^2 \sigma$$
$$\frac{\partial(\Delta F)}{\partial R} = 8\pi R\sigma - 4\pi R^2(f_1 - f_2)$$

If $\frac{\partial(\Delta F)}{\partial R} < 0$, then $R > \frac{2\sigma}{f_1 - f_2} = R_c$. That means when bubble is big, it will grow bigger to get lower free energy, when bubble is small, it will shrink to get lower free energy.